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water is one of the most important resources in the world, water is not a free commodity. it is the result of a very complicated series of natural processes, resulting in one of the most important geological processes of the earths surface. pollution of water is one of the most important contributors to water quality problems. in this review, we will focus on the contaminants that are found in water such as heavy metals, organic compounds, and radioactive ions. the source of the pollutants found in water can be the natural causes such as industrial wastewater, agricultural runoff, motor vehicle exhaust, power plant exhaust, and other human activities such as domestic and commercial discharges, as well as solid waste discharges and urban runoff. contaminants enter the aquatic environment directly through discharge of untreated or treated effluent, or can enter the aquatic

environment through soil erosion, run-off of fertilizers and pesticides, runoff of the soluble fraction of sediment, leaching from landfills, and point and non-point sources of effluent discharge. the sources

sources of effluent discharge. the sources of these pollutants to the ecosystem can be grouped into three main categories: natural sources, human sources, and industrial sources. in the body, the toxicity of these elements is due to their ability to form complexes with biological molecules, inhibiting metabolic enzymes, provoking oxidative stress, and decreasing the antioxidant defense of the organism. damage to essential enzymes and transcription factors mediated by xenobiotics can lead to disruption of cellular signal transduction cascades, culminating in inhibition of dna synthesis, cell cycle arrest, and apoptosis. general and specific toxicity are important factors in determining the risk to health associated with increased exposure to

metal elements, and the risks will vary between elements and within species. an understanding of toxicity is essential in developing strategies for limiting the effects of pollutants on the health of humans and wildlife.

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Tannins are plant-derived compounds that react with proteins and amino acids to form stable complexes or complexes with nitrogen-free radicals. Tannins can also be adsorbed by plant roots or cell walls and are transported to where they are needed in the plant (de Groot and de Vries 2014). In addition, they can also prevent phytoextraction of metals (Raab & Windfuhr 1997). Tannins and tanninbinding compounds, such as tannic acid and gallic acid, are found in all food plants, thus making it hard to develop techniques for the removal of tannins

from plant-derived foods (Raab & Windfuhr 1997). Freeze-dried processed products, such as gum arabic, are acidic in nature (with a pH of 3.8 - 4.5) (Hessemann 2010). Their stabilisation (reducing the likelihood of microbial growth) is achieved through the inclusion of salts (sodium chloride, potassium chloride or magnesium sulphate). At all stages of the papermaking process and especially when the objective is paper mill reclamation, a closer look at the chemistry and economics is needed to identify the best possible options. Although, it is common to use technology such as a thickening agent and/or "flocculant" (e.g. sodium alginate, starch, etc.) to improve pulp drainage, the amylopectin system is not completely understood. This reaction is non-specific and therefore may only be used to create lower quality paper, as it could potentially decompose into free lignin and lower the

strength of the sheet (Boutafina 2015 1392). Optimizing the enzymatic pulp drainage allows for the use of a cheaper, non-specific method that can be used to increase the brightness of the pulp (e.g. alkaline phosphatase), thus allowing for

the use of wet-end treatments that otherwise would be discarded. However, at low concentrations, the enzymes can degrade paper properties, especially strength (Boutafina 2015 1393). 5ec8ef588b

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